

New method to examine batteries -- MRI from the inside

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There is an ever-increasing need for advanced batteries for portable electronics, such as phones, cameras, and music players, but also to power electric vehicles and to facilitate the distribution and storage of energy derived from renewable energy sources. But, once a battery fails, there are no corrective measures—how do you look inside a battery without destroying it?

Now, researchers at Cambridge University, Stony Brook University, and New York University have developed methodology, based on <u>magnetic</u> <u>resonance imaging</u> (MRI), to do just that. Their technique, which is described in the journal *Nature Materials*, also creates the possibility of improving <u>battery</u> performance and safety by serving as a diagnostic of its internal workings.

MRI has been extremely successful in the medical field for visualizing disorders and assessing the body's response to therapy. However, MRI is not typically used in the presence of a lot of metal, a primary component in many batteries. This is because conducting surfaces effectively block the radio frequency fields that are used in MRI to see beneath surfaces or inside the human body.

The researchers, however, turned this limitation into a virtue. Because radio frequency fields do not penetrate metals, one can actually perform very sensitive measurements on the surfaces of the conductors. In the case of the popular lithium-ion batteries, for example, the team was able to directly visualize the build-up of lithium metal deposits on the



electrodes after charging the battery. Such deposits can also detach from the surface, eventually leading to overheating, battery failure, and – in some cases – to fire or explosion.

Visualizing small changes on the surface of the batteries' electrodes allows, in principle, for the testing of many different battery designs and materials under normal operating conditions.

The work is the result of a collaboration between Clare Grey, associate director of the Northeastern Center for Chemical Energy <u>Storage</u> and a professor at Cambridge and Stony Brook universities, and Alexej Jerschow, a professor in the Department of Chemistry at New York University who heads a multi-disciplinary MRI research laboratory.

"New electrode and electrolyte materials are constantly being developed, and this non-invasive MRI technology could provide insights into the microscopic processes inside batteries, which hold the key to eventually making batteries lighter, safer, and more versatile," said Jerschow. "Both electrolyte and electrode surfaces can be visualized with this technique, thus providing a comprehensive picture of the batteries' performancelimiting processes."

"MRI is exciting because we are able to identify where the chemical species inside the battery are located without having to take the battery apart, a procedure which to some degree defeats the purpose," added Grey. "The work clearly shows how we can use the method to identify where lithium deposits form on metal electrodes. The resolution is not yet where we want it to be and we would like to extend the method to much larger batteries, but the information that we were able to get from these measurements is unprecedented."

The project's other researchers were: S. Chandrashekar, a postdoctoral fellow at both Stony Brook and New York Universities, Nicole Trease, a



postdoctoral fellow at Stony Brook University, and Hee Jung Chang, a Stony Brook University graduate student.

"We still have some way to go to make the images better resolved, and make imaging time shorter," Chandrashekar noted, "We feel that with this work, we have made the field wide open for interesting applications."

The research team also envisions that the method could lead to the study of irregularities and cracks on conducting surfaces in the materials sciences field. In addition, they add, the methods developed here could be highly valuable in the quest for enhanced battery performance and in the evaluation of other electrochemical devices, such as fuel cells.

Provided by University of Cambridge

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