

## Uptake protein acts as zinc's doorway to the cell

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The cover art illustration of zinc ions (spheres) in a zinc-selective channel (tunnel) for the December 10, 2010, issue of the Journal of Biological Chemistry, in which the research is published. The inward spiral represents the gradient that drives zinc ions across the membrane, while the tetrahedral shape represents the specially shaped protein gateway that allows zinc ions to pass.

(PhysOrg.com) -- A study to be published as the "Paper of the Week" in the *Journal of Biological Chemistry* this December details how zinc, an element fundamental to cell growth, enters the cell via zinc-specific uptake proteins. The research, conducted at the U.S. Department of Energy's (DOE) Brookhaven National Laboratory, is the first to purify this kind of protein and study its role in zinc uptake.



Zinc is crucial to the health of all <u>living organisms</u>. At the cellular level, zinc is responsible for cell growth, which in turn affects the health, growth, and <u>reproduction</u> of an organism.

While there are six classes of known proteins that act as transporters or channels enabling zinc to cross the cell membrane, scientists have identified one metal-specific family of proteins whose purpose is to facilitate the cell's zinc uptake. These are called ZIP proteins, a reference to their resemblance to zinc-regulated and ion-regulated transporter proteins.

The exact mechanism by which ZIP proteins facilitate zinc uptake has been a mystery. Brookhaven scientists led by biologist Dax Fu — who have characterized other zinc-specific proteins that maintain healthy levels of zinc on either side of the cell membrane — have now taken a closer look at this process.

A long-held belief has posited that ZIP proteins work like elevators, pumping zinc across the cell membrane and into the cell. However, Fu and his colleagues found no evidence to support this explanation.

"This was a big surprise. For the last fifteen years, the assumption has been that the ZIP <u>protein</u> acts like a pump or elevator," said Fu. "Instead, we have found that ZIP is more like a door."

Fu and his colleagues have studied a ZIP protein provided by the New York Consortium on Membrane Protein Structure and derived from the bacteria Bordatella bronchiseptica. They expressed the protein in Eschericia coli, a bacteria whose zinc regulation has been well documented. Brookhaven's Jin Chai purified and concentrated the samples before exposing them to zinc. By reconstituting the purified ZIP proteins in this controlled manner, the scientists ensured that their sample would reflect only the ZIP mechanism for cellular zinc uptake.



"It's important to note that people have been trying to purify these proteins for a long time. Purification of the first ZIP family member opens the door for detailed structural and functional analysis at the molecular level," Fu said.

Using a fluorescent indicator, Brookhaven's Wei Lin then conducted several measurements to characterize zinc uptake, with attention to changes in zinc concentration, temperature, acidity, and electric charge.

The Brookhaven team found evidence of electrodiffusion. Ions diffuse by moving from a region with a high concentration to one of a lower concentration — like diners who relocate from a crowded dining hall to an adjoining, empty coffee room. In electrodiffusion, the diffused ions also change the electric charge of the space that they occupy. The imbalance in charge created by zinc ions moving into the cell builds during zinc uptake and acts against the concentration gradient, eventually causing zinc uptake to stop.

Based in part on the studies of similar metal-specific proteins, Fu and his colleagues have postulated that the ZIP protein allows zinc ion diffusion by providing an opening that is specifically shaped for zinc coordination chemistry. This hypothesis will eventually be confirmed in studies that crystallize and examine the ZIP protein at the atomic level.

"We are driven by our curiosity — we want to know how this works," Fu said.

Aside from satisfying scientific curiosity, this understanding could have a big impact. Zinc uptake at the cellular level is implicated in a range of biomedical and energy research. For example, in green plants, carbonic acid is converted to  $CO_2$  in a chemical reaction that is catalyzed by a zinc enzyme.



Zinc deficiency, therefore, has a direct impact on the carbohydrate metabolism of plants. For researchers developing biofuels energy sources, these systems and the role of zinc transporters in the conversion of energy into carbohydrates, are important objects of study. Developing a better understanding of <u>zinc</u> uptake can provide greater insight into these processes and will inform future discoveries.

**More information:** Scientific paper: <u>www.jbc.org/content/early/2010</u> ... 9/28/jbc.M110.180620

## Provided by Brookhaven National Laboratory

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