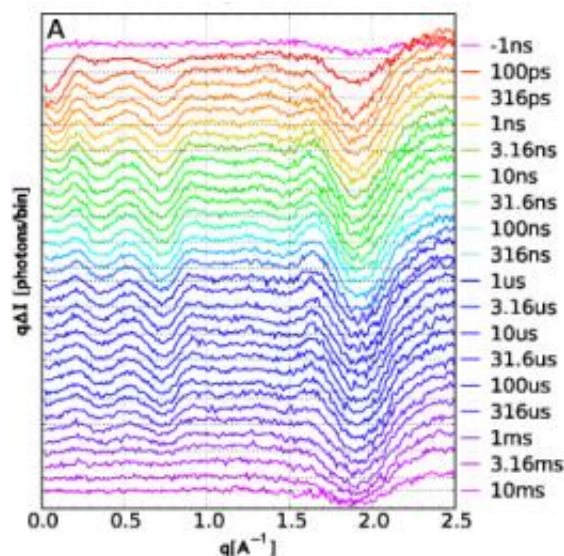


# Quick-Change Molecules Caught in the Act

June 1 2010



Structure change upon release of CO from myoglobin active site: Intensity of scattered X-rays as a function of distance from the center of the scattering pattern, for increasing intervals between laser and x-ray pulses. The shortest interval is at the top of the graph. The leftmost feature in the graph (SAXS region) is sensitive to the molecule's overall shape; features to the right of this (WAXS region) provide more detailed information about the protein structure. Colors are associated with proposed intermediate structures that arise in the course of the transition. (Copyright © 2010 by the National Academy of Sciences)

(PhysOrg.com) -- The chemistry of life happens so fast that a millionth of a second is an eternity -- an eternity that is largely invisible to science. In that time, molecules change in ways we cannot see. Now, though, there is a way of looking at molecules in solution -- as they are in life --

that shows both the big-picture shape and smaller-scale details of a molecule in one glance, with each glimpse being captured in one hundred trillionths of a second. With this technique, new chemical landscapes are now open to study.

Working at the BioCARS beamline 14-ID at the U.S. Department of Energy's [Advanced Photon Source](#) at Argonne National Laboratory, researchers from the National Institutes of Health and The University of Chicago have combined simultaneous small-angle and wide-angle x-ray scattering (SAXS and WAXS) with an ultrafast [laser](#) to observe the dynamics of a protein in solution with 100-picosecond time resolution and 2.5-Angstrom spatial resolution. The team demonstrated this unprecedented capability in a study of myoglobin.

Myoglobin is the protein that stores oxygen in muscles. The researchers looked at a reversible transition in which a carbon monoxide (CO) molecule binds to and releases from the active site in myoglobin. (The CO molecule is a stand-in for an oxygen molecule: They are chemically similar but the CO reaction is easier to observe with this technique.) The structures with and without CO are known, but are nearly identical. How, then, does the molecule change as CO gets out?

In the experiment, the release of CO was triggered by a pulse of laser light. At a given time delay ranging from 100 picoseconds to 10 milliseconds after the laser pulse, the sample was exposed to an x-ray pulse (see figure). X-rays that scatter at small angles (SAXS), tell about the overall shape of the myoglobin molecule. X-rays scattered at wide angles away from the sample (WAXS) give more detailed information about the structure.

The SAXS data show that the molecule immediately expands, but after 10 nanoseconds have passed, it's back to nearly normal size. In addition, signatures in the scattering data are consistent with the presence of

structures different from the original, and these signatures evolve over time. And because of the high spatial resolution of this technique, it may also be able to track when the CO molecule moves within the myoglobin structure and when it is excreted from the protein.

The method is made possible by a unique laser “pump-probe” setup developed in collaboration with NIH scientists as part of recent upgrade to beamline 14-ID, which is operated by BioCARS, part of the University of Chicago Center for Advanced Radiation Sources. The setup features a high-power, broadly tunable, short pulse laser system combined with sophisticated capabilities for timing the arrival of the laser pulse in relation to the x-ray pulse. In addition, given the design of the scattering diffractometer, the x-ray detector is large enough to capture both small- and wide-angle scattering without the need to reposition the detector.

Because this technique allows researchers to study [molecules](#) in solution rather than in single crystals, to collect SAXS and WAXS data simultaneously, to probe at the [picosecond](#) time scale, and to use light as a trigger, it is now possible to ask entirely new questions about a wide range of chemical and biological systems.

**More information:** Hyun Sun Cho, “Protein Structural Dynamics in Solution Unveiled via 100-ps Time-resolved X-ray Scattering,” PNAS, 107 (16), 7281 (20 April 2010). [DOI:10.1073/pnas.1002951107](https://doi.org/10.1073/pnas.1002951107)

Provided by Argonne National Laboratory

Citation: Quick-Change Molecules Caught in the Act (2010, June 1) retrieved 4 May 2024 from <https://phys.org/news/2010-06-quick-change-molecules-caught.html>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.