

## **'Cold' Mars Could Have Harbored Liquid** Water

June 1 2009, by Ruth Dasso Marlaire



Mars, as seen by the Hubble Space Telescope. Image credit: NASA

(PhysOrg.com) -- A new NASA study provides further evidence that Martian minerals dissolved in water could have kept that water from freezing, even on a cold, early Mars.

NASA scientists modeled freezing conditions on <u>Mars</u> to test whether liquid <u>water</u> could have been present to form the surface features of the Martian landscape.

Researchers report that fluids loaded with dissolved minerals containing elements such as silicon, iron, magnesium, <u>potassium</u> and <u>aluminum</u>, can remain in a liquid state at temperatures well below freezing. The results of this research appear in the May 21 issue of *Nature* magazine entitled "Stability Against Freezing of Aqueous Solutions on Early Mars."



"We found that the salts in water solutions can reduce the melting point of water, which may help explain how <u>liquid water</u> existed in a frozen Martian environment," said Alberto Fairén, a space scientist at NASA Ames Research Center, Moffett Field, Calif. and the lead author of the study.

To understand what formed the surface features on Mars, scientists have focused on the early Martian conditions. Was early Mars warm and wet, or cold and dry? Surface features throughout most of the Martian landscape suggest the presence of water ponds ranging from seas to lakes, and rivers and gullies formed by flowing water, which imply that early Mars was wet.

But there also is some evidence that suggests that Mars may have been permanently cold, with global temperatures well below the freezing point of pure water. To study the 'liquidity' of water on Mars, climate modelers first simulated various concentrations of greenhouse gases in its atmosphere. They found that these gases cannot efficiently raise the surface temperature above freezing.

A greenhouse atmosphere produced by carbon dioxide and water would have been saturated well below freezing. In addition, the amount of methane needed to raise the surface temperature above freezing, implies the planet had a terrestrial-like biological source for its methane supply, according to previous investigations.

Scientists then took another approach and looked at water solutions containing weathering basalts, similar to those seen at the Mars landing sites. They calculated these fluids' freezing points and evaporative processes. Results showed that a significant amount of weathering fluids containing silicon, iron, <u>magnesium</u>, calcium, chloride, sodium, potassium and aluminum remain in the liquid at temperatures well below freezing.



In addition, they studied the minerals that precipitated in the liquid solutions over time. These minerals are similar to those actually found on the Martian surface. Scientists concluded that salty liquid water on Mars may explain the stability of fluids against freezing on the Martian surface at temperatures below  $0^{\circ}$ C.

"Our goal was to learn how a combination of different processes of evaporation and freezing affect the freezing point of a hypothetical Martian solution. We also wanted to see how the liquid phases formed and destabilized over the evolution of different solutions," added Alfonso Davila, a co-author of the paper at NASA Ames Research Center, Moffett Field, Calif.

Provided by JPL/NASA (<u>news</u> : <u>web</u>)

Citation: 'Cold' Mars Could Have Harbored Liquid Water (2009, June 1) retrieved 18 April 2024 from <u>https://phys.org/news/2009-06-cold-mars-harbored-liquid.html</u>

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