

# Simulated gene therapy

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In a recent issue of *The Journal of Chemical Physics*, published by the American Institute of Physics (AIP), a group of researchers at the University of California, Berkeley and Los Alamos National Laboratory describe the first comprehensive, molecular-level numerical study of gene therapy. Their work should help scientists design new experimental gene therapies and possibly solve some of the problems associated with this promising technique.

"There are several barriers to gene delivery," says Nikolaos Voulgarakis of Berkeley, the lead author on the paper. "The genetic material must be protected during transit to a cell, it must pass into a cell, it must survive the cell's defense mechanisms, and it must enter into the cell's guarded nucleus."

If all of these barriers can be overcome, gene therapy would be a valuable technique with profound clinical implications. It has the potential to correct a number of human diseases that result from specific genes in a person's DNA makeup not functioning properly -- or at all. Gene therapy would provide a mechanism to replace these specific genes, swapping out the bad for the good. If doctors could safely do this, they could treat or even cure diseases like [cystic fibrosis](#), certain types of cancer, sickle cell anemia, and a number of rare genetic disorders.

Safety is a primary concern when working with gene therapy. Some of the first attempts at gene therapy used viruses to insert DNA into cells -- something that viruses naturally do anyway. Viruses can be dangerously toxic, however, and this fact was tragically demonstrated a decade ago

when an 18-year-old boy enrolled in a gene therapy study had a massive [immune reaction](#) to the viruses used. He died just a few days into the treatment from multiple organ failure, precipitating an immediate halt to the trial.

Since then, many alternatives to viruses have emerged for use in gene therapy, including [synthetic molecules](#) like "dendrimers," a word that derives from the Greek word for "tree." Similar to trees, dendrimers are branching molecules that are slightly positively charged. This allows them to be loaded with DNA (which is slightly negative charged) for insertion into a cell.

Dendrimers seem to offer many advantages over viruses. They may be much less toxic, and they may offer other advantages in terms of cost, ease of production, and the ability to transport very long genes. If they can be designed to efficiently -- and safely -- shuttle genes into human cells, then they may be a more practical solution to [gene therapy](#) than viruses.

So far, laboratory experiments with different types of dendrimers have shown that they can insert genes into cells, but only with very low efficiency. Hoping to discover the key to improving this efficiency, Voulgarakis and his colleagues simulated the detailed, atomic-level physical process of dendrimers entering cells. They varied parameters like the dendrimer size and the length of the DNA they carry. Modeling these parameters on a computer is a fast, inexpensive approach for testing different ideas and optimizing the delivery vehicle.

What they uncovered were the key factors that determine the success of dendrimers as gene delivery vehicles -- things like the charges of the dendrimers and their target cell membranes, the length of DNA, and the concentration of surrounding salt. Their work has illuminated some of the molecular-level details that should help clinicians design the most

appropriate gene vectors.

"Our study indicates that, over a broad range of biological conditions, the dendrimer/nucleic acid package will be stable enough to remain on the surface of the cell until translocation," says Voulgarakis.

Dendrimers are also used clinically for delivering cancer drugs to tumors, and for helping to image the human body. In the future, Voulgarakis and his colleagues plan to study the possibility of using dendrimers as drug delivery vehicles.

More information: The article " Dendrimers as Synthetic Gene Vectors: Cell Membrane Attachment" by N. K. Voulgarakis, K. Ř. Rasmussen, and P. M. Welch was published in the April 21, 2009 issue of *The Journal of Chemical Physics* [J. Chem. Phys. 130, 155101 (2009)]. See: [link.aip.org/link/?JCPSA6/130/155101/1](http://link.aip.org/link/?JCPSA6/130/155101/1) .

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