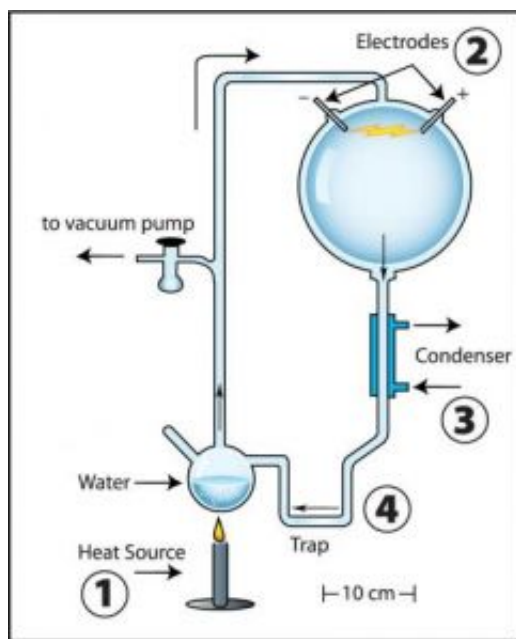


# Volcanoes May Have Provided Sparks and Chemistry for First Life

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A.

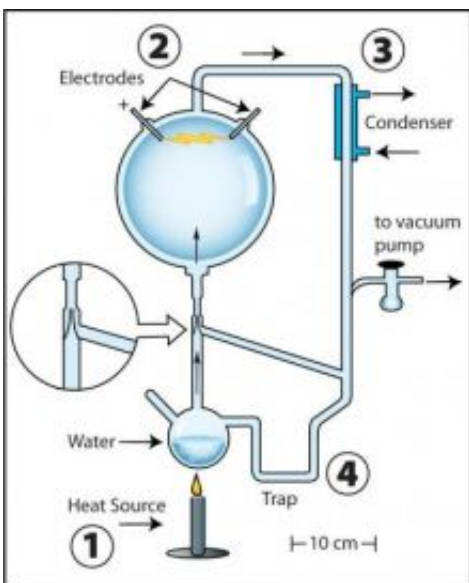
The apparatus used for Miller's original experiment. Boiled water (1) creates airflow, driving steam and gases through a spark (2). A cooling condenser (3) turns some steam back into liquid water, which drips down into the trap (4), where chemical products also settle. Credit: Ned Shaw, Indiana University

(PhysOrg.com) -- Lightning and gases from volcanic eruptions could have given rise to the first life on Earth, according to a new analysis of samples from a classic origin-of-life experiment performed more than fifty years ago. Using modern techniques to examine samples from the

original experiments, researchers discovered previously undetectable organic compounds.

A classic experiment proving amino acids are created when inorganic molecules are exposed to electricity isn't the whole story, it turns out. The 1953 Miller-Urey Synthesis had two sibling studies, neither of which was published. Vials containing the products from those experiments were recently recovered and reanalyzed using modern technology. The results are reported in this week's *Science*.

One of the unpublished experiments by American chemist Stanley Miller (under his University of Chicago mentor, Nobelist Harold Urey) actually produced a wider variety of organic molecules than the experiment that made Miller famous. The difference between the two experiments is small -- the unpublished experiment used a tapering glass "aspirator" that simply increased air flow through a hollow, air-tight glass device. Increased air flow creates a more dynamic reaction vessel, or "vapor-rich volcanic" conditions, according to the present report's authors.



B.

The apparatus used for Miller's "second," initially unpublished experiment. Boiled water (1) creates airflow, driving steam and gases through a spark (2). A tapering of the glass apparatus (inlay) creates a spigot effect, increasing air flow. A cooling condenser (3) turns some steam back into liquid water, which drips down into the trap (4), where chemical products also settle. Credit: Ned Shaw, Indiana University

"The apparatus Stanley Miller paid the least attention to gave the most exciting results," said Adam Johnson, lead author of the Science report. "We suspect part of the reason for this was that he did not have the analytical tools we have today, so he would have missed a lot."

Johnson is a doctoral student in IU Bloomington's Biochemistry Program. His advisor is biogeochemist Lisa Pratt, professor of geological sciences and the director of NASA's Indiana-Princeton-Tennessee Astrobiology Institute.

In his May 15, 1953, article in Science, "A Production of Amino Acids Under Possible Primitive Earth Conditions," Miller identified just five amino acids: aspartic acid, glycine, alpha-amino-butyric acid, and two versions of alanine. Aspartic acid, glycine and alanine are common constituents of natural proteins. Miller relied on a blotting technique to identify the organic molecules he'd created -- primitive laboratory conditions by today's standards. In a 1955 Journal of the American Chemical Society paper, Miller identified other compounds, such as carboxylic and hydroxy acids. But he would not have been able to identify anything present at very low levels.

Johnson, Scripps Institution of Oceanography marine chemist Jeffrey Bada (the present Science paper's principal investigator), National Autonomous University of Mexico biologist Antonio Lazcano, Carnegie

Institution of Washington chemist James Cleaves, and NASA Goddard Space Flight Center astrobiologists Jason Dworkin and Daniel Glavin examined vials left over from Miller's experiments of the early 1950s. Vials associated with the original, published experiment contained far more organic molecules than Stanley Miller realized -- 14 amino acids and five amines. The 11 vials scientists recovered from the unpublished aspirator experiment, however, produced 22 amino acids and the same five amines at yields comparable to the original experiment.

"We believed there was more to be learned from Miller's original experiment," Bada said. "We found that in comparison to his design everyone is familiar with from textbooks, the volcanic apparatus produces a wider variety of compounds."

Johnson added, "Many of these other amino acids have hydroxyl groups attached to them, meaning they'd be more reactive and more likely to create totally new molecules, given enough time."

The results of the revisited experiment delight but also perplex.

What is driving the second experiment's molecular diversity? And why didn't Miller publish the results of the second experiment?

A possible answer to the first question may be the increased flow rate itself, Johnson explained. "Removing newly formed molecules from the spark by increasing flow rate seems crucial," he said. "It's possible the jet of steam pushes newly synthesized molecules out of the spark discharge before additional reactions turn them into something less interesting. Another thought is that simply having more water present in the reaction allows a wider variety of reactions to occur."

An answer to the second question is relegated to speculation -- Miller, still a hero to many scientists, succumbed to a weak heart in 2007.

Johnson says he and Bada suspect Miller wasn't impressed with the experiment two's results, instead opting to report the results of a simpler experiment to the editors at Science.

Miller's third, also unpublished, experiment used an apparatus that had an aspirator but used a "silent" discharge. This third device appears to have produced a lower diversity of organic molecules.

Research on early planetary geochemistry and the origins of life isn't limited to Earth studies. As humans explore the Solar System, investigations of past or present extra-terrestrial life are inevitable. Recent speculations have centered on Mars, whose polar areas are now known to possess water ice, but other candidates include Jupiter's moon Europa and Saturn's moon Enceladus, both of which are covered in water ice. The NASA Astrobiology Institute, which supports these investigations, has taken a keen interest in the revisiting of the Miller-Urey Synthesis.

"This research is both a link to the experimental foundations of astrobiology as well as an exciting result leading toward greater understanding of how life might have arisen on Earth," said Carl Pilcher, director of the NASA Astrobiology Institute, headquartered at NASA Ames Research Center in Mountain View, Calif.

Provided by Indiana University

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