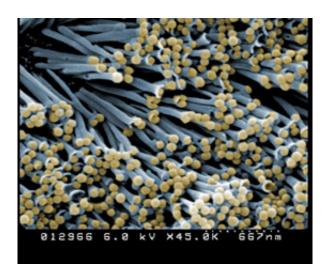


World's tiniest test tubes get teensiest corks

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Rows of tiny nano test tubes rest on a mesh in this electron microscope photo, shot in October of last year and colorized for added clarity. Seeking to deliver drugs within the body, University of Florida scientists have found a way to "cork" the tubes. The goal is to develop the ability to fill the teeny tubes with drugs and inject them into the body, where they will seek diseased or cancerous cells, uncork and spill their therapeutic contents in the right place. Credit: University of Florida

Now all they need is a really, really small corkscrew. Like Lilliputian chemists, scientists have found a way to "cork" infinitesimally small nano test tubes. The goal is a better way to deliver drugs, for example, for cancer treatment. Scientists want to fill the teeny tubes with drugs and inject them into the body, where they will seek diseased or cancerous cells, uncork and spill their therapeutic contents in the right place.



"After making the nano test tubes, we saw the potential for them to be used for drug delivery vehicles, but because they are open at one end it would be like trying to ship wine in a bottle without a cork," said University of Florida chemistry professor Charles Martin. "You have to cork it, which is what we have accomplished."

Martin is one of six University of Florida chemistry faculty members and graduate students who co-authored a paper about the research that appeared last month in the *Journal of the American Chemical Society*.

While chemotherapy works against many cancers, it can cause severe side effects such as nausea, temporary hair loss and blood disease. To make the chemo hit only the cancerous cells, Martin and scientists elsewhere have spent recent years experimenting with drug-carrying nanotubes or nanoparticles.

"Nano" stems from nanotechnology, the fast-growing science of making objects or devices that approach molecular dimensions. One nanometer equals one-billionth of a meter.

The approach makes sense for attacking diseased cells while bypassing healthy ones, but it also poses challenges. For one thing, the nanotubes must recognize their target, a problem scientists are attacking by tweaking their chemistry to make it respond to the unique chemistry of cancer cells. The tubes also must be biologically benign. Martin says a method for making nanotubes he pioneered, template synthesis, allows manufacturers to use biodegradable material, such as the polylactides that compose biodegradable sutures.

Additionally, the tubes also had to be closed at one end to form the classic test tube shape, a problem Martin and his group solved in research published in 2004.



To "cork" the test tubes in the latest research, the researchers applied an amino chemical group to the mouth of the tubes and an aldehyde chemical group to the corks. The two groups are complementary, so they bond with one another.

Billions of nanotubes could fit on a postage stamp. So, said Martin, "we don't put individual caps in each nanotube the way corking machines do for bottles."

Instead, the scientists immerse a small mesh that holds millions of aminomodified nanotubes, all precisely lined up in a grid pattern, into a solution imbued with millions of the corks. Brownian motion — what happens when minute particles immersed in a fluid move about randomly — takes care of the rest. The corks simply float around, then slip into the mouths of the tubes as they encounter them.

The diameter of the tubes is about 80 nanometers, or 80-billionths of a meter. Even though they are tiny, each tube can hold about 5 million drug molecules. "Each tube packs a real punch in terms of the number of drug molecules it can deliver," Martin said.

Sang Bok Lee, an assistant professor of chemistry and biochemistry at the University of Maryland, works on similar research. He said scientists have proposed capping the tubes using chemical interactions between the drugs and the tubes. But that might not work because the tube could leak before it reaches its target.

"I strongly agree that Professor Martin's proposed strategy will be one of the ideal solutions for the problem of controlling drug uptake and release," he said in an e-mail.

The UF scientists aren't there yet. There's no easy way to unlock the amino chemical group from the aldehyde chemical group. So while



Martin says there are some promising possibilities, he and his colleagues have their next job cut out for them: figuring out how to uncork the tubes.

Source: University of Florida

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