

# Zinc Powder Will Drive Your Hydrogen Car

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A revolutionary method of using concentrated solar energy for producing hydrogen in a clean, safe and inexpensive way was developed by a cooperation of scientists from Israel, Sweden, Switzerland and France. This new method, based on the production of pure zinc, may enable an easier and quicker move to a hydrogen-based economy thus reducing the need for the depleting petroleum and its polluting side effects.

*Image: The largest computerized mirrors array in the world, located in the Israeli Weitzman Institute. The array is capable of concentrating the sun energy to the desired location, achieving extremely high temperatures.*

The recent energy crisis served to emphasize the urgent need for a replacement for fossil fuels. The best known candidate to replace

petroleum as the main fuel of the 21st Century automotive industry is hydrogen. Hydrogen can be found in great abundance (mainly as part of water); it does not pollute the air as it burns and it is more energetic than most existing fuels. The main reason this promising substitute for petroleum has not reached the main market is the staggering costs of producing and transporting it.

As mentioned, the main source for hydrogen production is water. Today, the most common method of production is electrolysis. Electrolysis decomposes the water molecules to its components (hydrogen and oxygen) by passing a strong electric current through it. This process is relatively simple but it requires a significant amount of electricity and thus is currently considered to be too expensive for large scale production of hydrogen.

Breaking the water molecules by simply heating them is not practical, since it requires achieving temperature above  $2,500^{\circ}\text{C}$  ( $4,530^{\circ}\text{F}$ ). Many years ago it was discovered that it is possible to use pure zinc to extract the oxygen from water, therefore releasing hydrogen. This process can be done in the much lower temperature of  $350^{\circ}\text{C}$  ( $662^{\circ}\text{F}$ ).

Since zinc is a relatively abundant metal, and is the fourth among all metals in world production – being exceeded only by iron, aluminum and copper – it can be considered a natural choice for producing hydrogen. The problem is that the current industrial production of pure zinc (Zn) from zinc oxide (ZnO) by either electrolysis or smelting furnaces is characterized by its high-energy consumption and concomitant pollution, derived mostly from the combustion of fossil fuels for heat and electricity.

To address the issue solar energy can be used as the main energetic source in the production of Zn from ZnO. In 2004 the European Union

and the Swiss Federal Office of Science and Education decided to fund a joint research to explore this possibility using a 45 kW solar furnace in Villigen, Switzerland, a 75kW solar simulator in Zurich and the largest solar research facility in the world with 1MW output located at the Weitzman Institute in Israel. The need for such a large amount of power is due to the need for very high temperatures required for the production of Zn from ZnO (normally around 1750°C / 3182F). Adding small amounts of carbon in the form of coal enabled the Weitzman team to reduce the Zn production temperature to a more manageable 1200°C (2192F). For the future, the team sees the possibility of replacing the coal completely with biomass thus making the entire process completely pollution free.

The real achievement of the Weitzman team is the scale of production of Zn which reached an average of about 50kg/h during tests using the existing Solzinc solar reactor located at the center of the Institute. On a full scale industrial facility much larger amounts could be extracted using a similar process. The energy cycle developed by the researchers is very efficient and relatively self-sustaining. ZnO is mined and transported to the Solzinc solar facility where it is mixed with small amounts of coal and put inside the solar furnace located on top of a high tower. A large array of heliostats (computer guided highly reflective mirrors) follows the sun around the sky and reflects the light to a hyperbolic mirror located inside the solar tower producing highly concentrated heat inside the solar furnace. At a heat of above 1200°C (2192F) the ZnO breaks down into Zn and oxygen which in turn recombines with the carbon to create CO as a minor by-product. The Zn is then cooled down to create a fine powder which can be safely handled and transported. In order to produce hydrogen from the Zn powder a much simpler process is performed where the Zn is mixed with water at a temperature of 350°C (662F). The oxygen inside the water recombines with the Zn to produce ZnO once again and the by-product is pure hydrogen.

This process has many advantages over existing ways of producing hydrogen. First and foremost it uses a renewable form of energy - the sun. Furthermore the main material required for the process is the relatively inexpensive zinc oxide which is almost completely recycled back by the end of the process. Another important advantage is that the hydrogen could be produced where it is needed, i.e., at the local fuel station instead of transporting large amounts of explosive hydrogen across the country. Trucks loaded with safe Zn powder would transport it to the fuel station where it would undergo the relatively simple treatment of extracting the hydrogen using steam. Last but not least, the entire process is relatively clean and when biomass will replace the coal as an additive to the ZnO mix, the process will be completely nonpolluting.

And now for the 10 million dollar question: if the process is so simple and efficient why can't you buy zinc-based hydrogen fuel at your local fuel station right now? First of all, there is the lack of infrastructure. Currently there are almost no hydrogen fuel stations, let alone hydrogen-based cars. This could change within 5-10 years but it will require great investments by both countries and the industry. It will also require large amounts of inexpensive accessible hydrogen, and here enters the Solzinc process developed at the Weizman Institute. As Michael Epstein, head of the Solar Research Facilities Unit in the Weizman Institute explained in an interview to IsraCast, it is now up to the industry to push the project forward into full commercialization. His assessment is that with the backing of the industry, full-scale production of hydrogen from Zn using the industrial Solzinc facility could begin within 8-10 years. In the meantime, Epstein is considering more immediate uses for Zn such as Zinc-Air batteries which are similar to existing batteries but get one of their main reactants-oxygen-from the outside air. These batteries are nontoxic and are neither highly reactive nor flammable. Several companies across the world are currently developing Zinc-Air batteries; among the leading ones is the Israeli company Source: [Electric Fuel](#) which develops Zinc-Air battery technology for automobiles.

*Pictures courtesy of the Weitzman Institute*

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